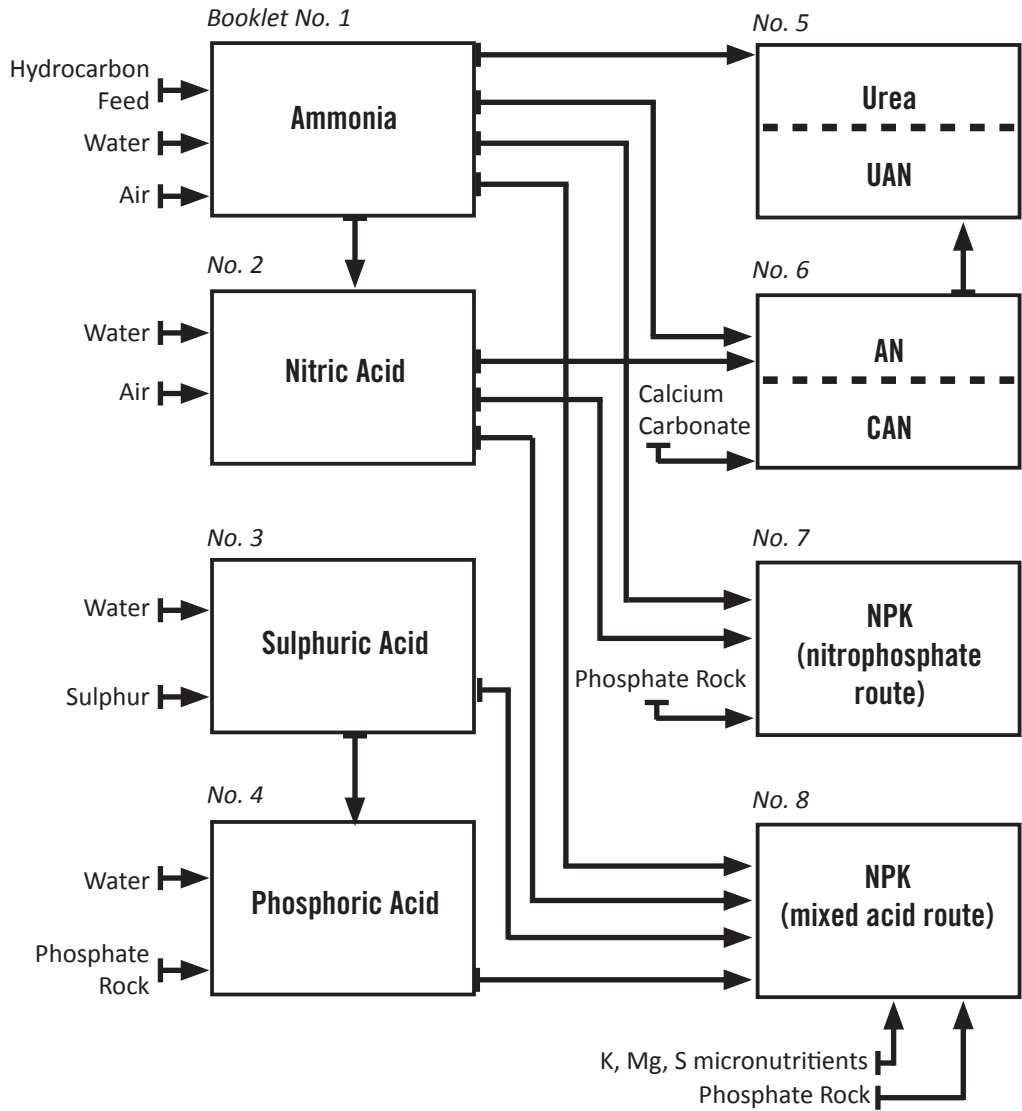




Fertilizers Europe BAT Booklets





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2. Calcium Ammonium Nitrate (CAN)

2.1 Identification

Chemical name	: Mixture of ammonium nitrate and calcium Carbonate Calcium ammonium nitrate, CAN
Composition	: Mixture of ammonium nitrate with calcium car- bonate and/or dolomite containing not more than 80% of ammonium nitrate

2.2 Hazards to Man and the Environment

To man

CAN is basically harmless when handled correctly.

To the environment

CAN is basically harmless when handled correctly.

2.3 Physical and Chemical Properties

Appearance	: White, off-white or grey granules or prills
Odour	: Odourless
pH water solution (10g/100ml)	: >4.5
Explosive properties	: Not explosive as per EEC test A14. The fertilizer has a very high resistance to detonation. This resis- tance is decreased by the presence of contaminants and/or high temperatures
Oxidising properties	: Not classified as oxidising material according to EEC Directive 88/379/EEC. Can support com- bustion
Solubility in water	: NH_4NO_3 highly soluble $\text{CaCO}_3/\text{MgCO}_3$ sparingly soluble
Bulk density	: $900\text{-}1,100\text{kg.m}^{-3}$











Best Available Techniques Booklets were first issued by Fertilizers Europe in 1995. Second revised edition 2000.

1. Production of Ammonia
2. Production of Nitric Acid
3. Production of Sulphuric Acid (in collaboration with ESA)
4. Production of Phosphoric Acid
5. Production of Urea and Urea-Ammonium Nitrate
6. Production of Ammonium Nitrate and Calcium Ammonium Nitrate
7. Production of NPK Compound Fertilizers by Nitrophosphate Route
8. Production of NPK Compound Fertilizers by Mixed Acid Route



Avenue E. van Nieuwenhuysse 4/6
B-1160, Brussels, Belgique
Tel: +32 2 675 3550
Fax: +32 2 675 3961
main@fertilizerseurope.com

www.fertilizerseurope.com

twitter.com/FertilizersEuro



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